The Influence of Chirality in the Amide Side Chain on the Carbonyl Orientation in Rotational Isomers of 3-Carbamoylpyridinium Halides

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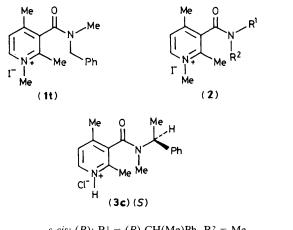
The direction of the carbonyl orientation in solid amide rotamers of $3-(N-methyl-N-\alpha-methylbenzylcarbamoyl)-$ 1,2,4-trimethylpyridinium iodide is governed by the (*R*)- or (*S*)-chirality in the amide side chain; *X*-ray structures and c.d. spectra are correlated.

In a previous communication¹ we reported the isolation of the amide rotamer (1t). In addition it was found by X-ray analysis and c.d. spectrometry that spontaneous resolution of the axially dissymmetric compound had occurred.[†] We then introduced additional carbon chirality in the amide side chain (arbitrarily the *R*-isomer) to study its effect on the carbonyl orientation with respect to the pyridinium ring. As for (1) we isolated rotamerically pure (2c) (*R*) and (2t) (*R*).

X-Ray analysis¹ revealed that in (2t) (R) the benzylic hydrogen is positioned towards the pyridinium ring. This implies a carbonyl orientation directed to the A-side[‡] of the

[†] Equilibration of amide rotamers and the chromatographic separation of the enantiomers have been observed in amides of unsymmetrically substituted mesitoic acids.²

 \ddagger The *A*-side of the pyridine ring is the side which faces the observer when the ring is viewed from a direction perpendicular to the plane of the ring and one travels around the ring in a counterclockwise direction when taking the shortest path from the ring N atom to the carboxamide group. The other side is the *B*-side.



s-cis; (R): $R^1 = (R)$ -CH(Me)Ph, $R^2 = Me$ (S): $R^1 = (S)$ -CH(Me)Ph, $R^2 = Me$ s-trans; (R): $R^1 = Me$, $R^2 = (R)$ -CH(Me)Ph (S): $R^1 = Me$, $R^2 = (S)$ -CH(Me)Ph

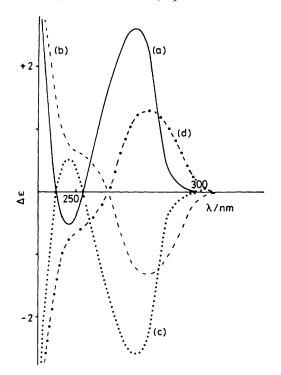


Figure 1. C.d. spectra (in MeOH) of: (a) (2t)(R); (b) (2c)(R); (c) (2t)(S), and (d) (2c)(S).

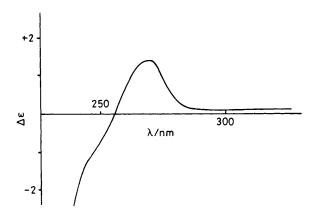


Figure 2. C.d. spectrum (in MeOH) of (3c) (S) measured on a Yobin-Yvon Mark III-S spectrometer.

pyridinium ring, according to the helicity rule³ designated as (M). C.d. spectra recorded for solutions of a single crystal and a larger sample were identical. Therefore we concluded that in compounds of type (2t) the carbonyl orientation is governed by the chirality in the amide side chain.

To substantiate this conclusion we have now repeated the amide synthesis¹ using the (S)-amine to produce (2t) (S). This compound, in comparison with (2t) (R), has an identical ¹H n.m.r. spectrum and indeed shows an opposite Cotton effect (see Figure 1). So, the (R) and (S) isomers may be regarded as enantiomers with opposite configurations. Analogous to the preparation of (2c) (R), which was briefly mentioned,¹ we

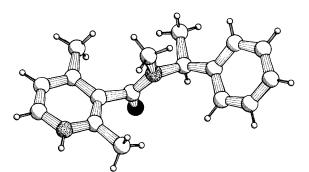


Figure 3. Perspective view of the absolute configuration of (3c) (S). Only one of the independent (nearly identical) molecules is shown. N atoms are indicated by dotted spheres and O by a black sphere.

isolated (2c) (S). As expected also in the s-cis-series both compounds show identical ¹H n.m.r. spectra but opposite Cotton effects (Figure 1). Attempts to obtain single crystals from (2c) (S) to elucidate its absolute structure have so far been unsuccessful. Therefore we prepared the HCl salt (3c) (S)§ directly from the parent pyridine, which is known to possess the s-cis-conformation.¹ The c.d. spectrum of (3c) (S) (see Figure 2) indicates the structural similarity with (2c) (S)

The result of the X-ray analysis¶ is represented in Figure 3 n which the carbonyl orientation is directed to the B-side‡ of he pyridinium ring resulting in a helicity designated as (P).³ Summarizing, we conclude that in compound (2) the absolute configuration in the amide side chain determines the absolute axial chirality which is not affected by the s-*cis*- or s-*trans*conformation of the amide moiety.

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References

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 $\$ Compound (3c) (S) was prepared by passing dry HCl gas through a solution of the parent pyridine in Et₂O: ¹H n.m.r. (D₂O) δ : 1.69 (3H, d, 9-Me), 2.70 (9H, s, 2-Me, 4-Me, N-Me), 6.08 (1H, q, 9-H), 7.55 (5H, s, Ph), 7.98 (1H, d, 5-H), 8.69 (1H, d, 6-H). Single crystals were obtained from PrⁱOH.

¶ Crystal data for (3c) (S): $C_{17}H_{21}CIN_2O \cdot H_2O$, M = 322.83, monoclinic, space group P_{21} , a = 13.667(1), b = 8.585(1), c = 16.106(4) Å, $\beta = 106.09(1)^\circ$, U = 1815.7(5) Å³, $D_c = 1.181$ g cm⁻³, Z = 4, F(000) = 688, Ni-filtered Cu- K_{α} radiation, $\lambda = 1.54184$ Å, $\mu(Cu-K_{\alpha}) = 19.4$ cm⁻¹. Intensities of 7232 reflections were collected (CAD4F; $2\theta_{max} = 140^\circ$; $\omega/2\theta$ scar; hkl/hkl Friedel pairs) of which 6021 with $I > 2.5\sigma(I)$ were used in the solution and refinement. The structure was solved by Patterson and Fourier methods and refined by least-squares techniques to R = 0.0659. H-atoms were placed at calculated positions with X–H = 0.98 Å. The alternative absolute configuration was rejected based on the refinement of the inverted model which resulted in the significantly higher R = 0.0776. Atomic co-ordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.